Kharagpur Lectures

Classical Molecular Dynamics and Computational Statistical Mechanics

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Lagrange's Formulation of Mechanics
Least-Action PrincipleImage: Constraint of the experiment of the experim





R.E. Gillilan and K.R. Wilson, J. Chem. Phys. 97 (1992).



gfortran : Free Fortran Compiler gnuplot : Free Computer Graphics Software

gfortran is a Fortran standard compiler. Downloads for the binaries are available for most systems.

Well documented and conforms to standards up to the year 2000. Trivial to install on Mac systems and Linux. Harder to install on Windows systems. Use Google search for *gfortran downloads*.

gnuplot is free graphics software used quite often for scientific research.

Well documented with a large collection of demonstration problems . Google search *gnuplot* will take you to the Home page.

Our uses : Publications and books, research projects

Graphics capability :

Line plots, point plots, contour plots, color labeling, multiple plots per page. Many advanced capabilities include3d hidden surfaces, transparency, and coloring of surfaces.





Fourth-Order Runge-Kutta Algorithm

Compute r,v by averaging 4 values of derivatives t = {0, dt/2, dt} :

$$r(dt) = r(0) + \frac{dt}{6}(v_1 + 2v_2 + 2v_3 + v_4)$$
$$v(dt) = v(0) + \frac{dt}{6}(a_1 + 2a_2 + 2a_3 + a_4)$$

$$r_{1} = r(0); \dot{r}_{1} = v_{1} = v(0); \dot{v}_{1} = a_{1}(r_{1});$$

$$r_{2}\left(\frac{dt}{2}\right) = r_{1} + \frac{dt}{2}v_{1}; v_{2}\left(\frac{dt}{2}\right) = v_{1} + \frac{dt}{2}a_{1}; a_{2} = \dot{v}_{2};$$

$$r_{3}\left(\frac{dt}{2}\right) = r_{1} + \frac{dt}{2}v_{2}; v_{3}\left(\frac{dt}{2}\right) = v_{1} + \frac{dt}{2}a_{2}; a_{3} = \dot{v}_{3};$$

$$r_{4}(dt) = r_{1} + v_{3}dt; v_{4}(dt) = v_{1} + \frac{dt}{2}a_{3}.$$





Fifth Order Runge-Kutta Integrator 6 force evaluations	
<pre>call fcn(yy,yyp) yak1 = yyp ynew = yy + (dt/2)*yak1</pre>	
<pre>call fcn(ynew,yyp) yak2 = yyp ynew = yy + (3*yak1 + yak2)*dt/16</pre>	
<pre>call fcn(ynew,yyp) yak3 = yyp(i) ynew = yy + yak3*dt/2</pre>	
<pre>call fcn(ynew,yyp) yak4 = yyp ynew = yy + (-3*yak2 + 6*yak3 + 9*yak4)*dt/16</pre>	
<pre>call fcn(ynew,yyp) yak5 = yyp ynew = yy + (yak1 + 4*yak2 + 6*yak3 - 12*yak4 + 8*yak5)*dt/7</pre>	
<pre>call fcn(ynew,yyp) yak6 = yyp</pre>	
<pre>yyp = (7*yak1 + 32*yak3 + 12*yak4 + 32*yak5 + 7*yak6)*dt/90 yy = yy + dt*yyp(i)</pre>	













Thermodynamics & Hydrodynamics from Dynamics

Equilibrium state variables: E, V;

 $\{F = ma\} \rightarrow T \text{ and } P$.

Dynamics: Compute P and T from :

$$(x, y, z)$$
, (v_x, v_y, v_z) , (F_x, F_y, F_z)

Define kinetic temperature as the mean squared values of the velocity components :

$$(kT/m) \equiv \langle v_x^2 \rangle = \langle v_y^2 \rangle = \langle v_z^2 \rangle.$$

Use the Virial Theorem for the Pressure Tensor P

Away from equilibrium :

Use the Heat Theorem for the Heat-Flux vector Q

The Virial Theorem for the Pressure Tensor P



Multiply x-component the pairwise force components by the coordinate :

$$\sum x_i \cdot F_{ij} = \sum x_{ij}^2 f_{ij} / r_{ij}$$

Notice also : $m(d/dt)\langle x_i\dot{x}_i\rangle = m\langle \dot{x}_i^2\rangle + m\langle x_i\ddot{x}_i\rangle$

Consider the x-component and time average over all particles and add wall forces.

$$\sum_{i} m \langle x_{i} \dot{x}_{i} \rangle = \sum_{ij} \langle x \cdot F_{ij} \rangle - P_{xx} V = \sum_{i} m (d/dt) \langle x_{i} \dot{x}_{i} \rangle - \sum_{i} m \langle \dot{x}_{i}^{2} \rangle$$

where $x_i \dot{x_i}$ is a bounded quantity and $NkT_{xx} = \sum_i m \langle \dot{x_i}^2 \rangle$ in two dimensions.

The long-time derivative of a bounded quantity is zero, giving the Virial Theorem .





Rudolf Clausius 1822-1888



4. Monte-Carlo Methods Two random number generators Monte-Carlo generation of the Maxwell Boltzmann distribution and moments Equilibrium Monte-Carlo Averages, Mayer cluster integrals, pair distribution and collision rate



```
function rund(intx,inty)
i = 1029*intx + 1731
j = i + 1029*inty + 507*intx - 1731
intx = mod(i,2048)
j = j + (i - intx)/2048
inty = mod(j,2048)
rund = (intx + 2048*inty)/4194304.0
return
end
```

Use the program "federico" in arxiv:1305.0961v2 ending the loops with 7 and write the seeds for 7 random numbers. Write another program to read the file of seeds and write out both forward and reverse numbers. Notice that the random number seeds are reversed when rund is called with the seeds lntx = 1862 and inty = 1347. The random numbers generated are :

1 0.67508363723754883 0.74579644203186035 2 0.16147541999816895 0.65815877914428711 3 0.90861988067626953 0.97026991844177246 4 0.97026991844177246 0.90861988067626953 5 0.65815877914428711 0.16147541999816895 6 0.74579644203186035 0.67508363723754883

How do you decide if this is a good random number generator?





Notice the odd moments vanish. Why?

Equation of State Averages from Statistical Mechanics Hard-Disk Monte Carlo

The virial expansion is a representation of the pressure in powers of the density :

$$(PV/NkT) - 1 = B_1\left(\frac{A_0}{A}\right) + B_2\left(\frac{A_0}{A}\right)^2 + B_3\left(\frac{A_0}{A}\right)^3 + \cdots \text{Hard disks}$$

$$B_2 = \pi/3^{1/2}$$
; $B_3 = 4\pi^2 A_{3,3}/9$; $B_4 = \pi^3 (6A_{4,5} - 3A_{4,4} - A_{4,6})/3^{3/2}$

The $A_{i,j}$ are cluster integrals and represent the volume in configuration space in which i particles form j bonds. The Mayer f functions describing the clusters are integrated over the volume and are proportional to the constants in the density expansion¹. In Reference [2] below a Modified Monte Carlo technique is described which uses interacting hard spheres to numerically compare with the expansion using cluster integrals developed by the Mayers.

- 1. J. E. Mayer M. G. Mayer, *Statistical Mechanics,* (John Wiley and Sons, Inc., New York, 1940) pp. 277-2911.
- 2. Nicholas Metropolis, Ariana W. Rosenbluth, Marshall N. Rosenbluth, Augusta Teller, and Edward Teller, "Equation of State Calculations by Fast Computing Machines,

Mayer Cluster integrals

The virial coefficients are proportional to the "star" cluster integrals developed by the Mayers. The Mayer derivation involves expanding the partition function in terms of the product of single particle distributions to form two, three, ..., n-particle distributions. These doubly-connected products represent *clusters* formed by bonds between the n-particles. Below are examples of the *star clusters* formed for 3, 4, and 5 particles. The subscripts (i, j) denote the number of particles and the number of Mayer f-bonds in the configuration. Each figure denotes a unique configuration.





Calculation of B₃ in Three Dimensions

```
implicit double precision (a-h,o-z)
top = 0
bot = 0
intx = 0
inty = 0
itmax = 10 000 000
iout = itmax/10
do it = 1,itmax
x3 = rund(intx, inty) * 2 - 1
y3 = rund(intx, inty) * 2 - 1
z3 = rund(intx, inty) * 2 - 1
x^2 = rund(intx, inty) * 2 - 1
y^2 = rund(intx, inty) * 2 - 1
z2 = rund(intx, inty) * 2 - 1
r12 = dsqrt(x2*x2 + y2*y2 + z2*z2)
r13 = dsqrt(x3*x3 + y3*y3 + z3*z3)
r23 = dsqrt((x2-x3)**2 + (y2-y3)**2 + (z2-z3)**2)
if((r12.lt.1).and.(r13.lt.1).and.(r23.lt.1)) top = top + 1
if((r12.lt.1).and.(r13.lt.1)) bot = bot + 1
if(iout*(it/iout).eq.it) write(6,*) top,bot,top/bot,15/32.0d00
enddo
stop
end
```

Equilibrium Monte-Carlo Method for Canonical Averages

It follows from statistical mechanics that phase-space averages calculated by weighting microstates with the Boltzmann factor will converge to the canonical-ensemble averages for a physical variable X(q,p):

$$\langle X \rangle = \Sigma X_i e^{-E_i/kT} / \Sigma e^{-E_i/kT}$$

For more than a few degrees of freedom random sampling of phase space will generate very few, if any at all, accepted microstates. This follows because the ratio of the occupied and the unoccupied part of phase space is close to zero. The Modified Monte Carlo Method of Metropolis, the Rosenbluths', and Tellers' corrects this deficiency for hard disks by examining microstates with relatively small energy changes, $\Delta \phi = kT$, in configuration space only. Then the microstate weights are not needed! The microstates are weighted evenly.

$$\langle X \rangle = \sum X_i / \sum 1$$

The detailed steps in the algorithm are listed on the next vugraph.



Results	for the Monte-Carlo Equation of	
State by	Metropolis, Rosenbluths, Teller	S

TABLE I. Results of this calculation for $(PA/NkT)-1=X_1$ compared to the free volume theory (X_2) and the four-term virial expansion (X_3) . Also (PA_0/NkT) from our calculations.

v	(A/A_0)	<i>X</i> 1	X2	X.	(PAo/NkT)	
2	1.04269	49.17	47.35	9.77	48.11	Solid phase
5	1.31966	6.43	6.72	5.35	5.63	Sond phase
5.5	1.4909 1.7962	4.41 2.929	4.53 2.939	4.02 2.680	3.63 2.187	
6.25	2.04616	2.186	2.323	2.065	1.557	
7	4.04145	0.6766	0.990	0.667	0.4149	

X₁ is the Monte-Carlo calculation.

 X_2 is the free volume estimate.

 X_3 is the four term virial series. Ten virial terms are known now.

The virial expansion only describes the fluid state. Today these numbers are accurate to five figures.



Pair Distribution FunctionThe pair distribution function was used in the early days of molecular dynamics
to compare potential models with neutron and xray diffraction results.Consider the ratio of pairs of particles with separation distance in a range dr .
The pair distribution is constructed with bins of length dr containing the ratio of
the number of pairs for a potential ϕ to the number of pairs for an ideal gas. We
consider two potentials, an 8-4 potential and a repulsive potential. $\Phi_{84} = (2 - r^2)^8 - 2(2 - r^2)^4$; $r_{cutoff} = \sqrt{2}$
 $\Phi_{repulsive} = 100(1 - r^2)^4$; $r_{cutoff} = 1.0$ $N_k^{\phi} = \# pairs(r_k \le r_c < r_{k+1})$
 $N_k^{ideal gas} = 2\pi r_c dr N(N-1)/2V = \pi r_c dr(N-1)\rho$ $N_k = \frac{N_k^{\phi}}{N_k^{ideal gas}} \longrightarrow g(r_k) = \frac{\sum N_k}{itmax}$ We use 400 bins of size 0.01 and max(r_c) = 4 to construct g(r) shown next.



A Calculation of Equilibrium Averages with the Center of Mass set to Zero

The center of mass motion should always be set to zero when periodic boundary conditions are used. The effect of this is to increase the collision rate and therefore the pressure. The Virial Theorem for hard disks (rearranged terms) is

$$\frac{PA}{NkT} - 1 = \left\langle \sum_{l=1}^{N} r_l \cdot F_l \right\rangle / m \sum_{l=1}^{N} v_l^2 = (Nu^2 \tau)^{-1} \sum_{collisions}^{ln \tau} r_c \cdot v_c$$

where the last term on the right is a collision term coming from the average over the internal pair forces and u^2 and τ are the mean squared velocity and the last sum measures the number of collisions occurring in a time τ .

Taking the ratio of the above equation with its low density form gives

$$g(\sigma) = \frac{PA/NKT - 1}{B_2(N/A)} = \frac{\Gamma}{\Gamma_0}$$

Here B_2 is the second Virial coefficient and the righthand side is the ratio of the high density to the low density collision rate. Both B_2 and Γ_0 are increased by the center of mass correction. The collision rates are calculated from molecular dynamics and agree with the calculation from the virial theorem as shown on the next vugraph.

	PA/NkT^{a}	collisions	Monte Carlo	Adjusted Monte Carlo	A/A0	
.00	21.9	2	20.62-4-0.09	22.40	1.100	
50	9.99	2	9.18 ± 0.06	9.93	1.250	·
50	8.10	20	7.42 ± 0.08	8.01	1 350	۰ ا
00	7.42	10	6.86 ± 0.08	7.39	1 400	
-50	6.93	10	6.27 ± 0.09	6.75	1.400	
75	6.61	20	6.21 ± 0.08	6.69	1.450	1
500	6.33	20	5.91 ± 0.05	6.36	1.475	1
525	6.17	20	5.79 ± 0.06	6.22	1 500	1
550	5.94	10	5.52 ± 0.06	5.94	1.500	12
600	5.56	10	5.23 ± 0.07	5.61	1.525	1
550	5.20	10		•••	1.550	1
700	4.80	7	4.66 ± 0.07	4.99	1.600	11
500	4.26	4	4.07 ± 0.06	4.34	1 700	
000	3.93	4	•••		1.700	
000	3.54	4	3.37 ± 0.06	3.59	1.800	1
100	3.30	2	•••		2.000	1
000	2.10	2	2.03 ± 0.03	2.12	3 000	1
A / A ₀ The ac	4.26 3.93 3.54 3.30 2.10 * These of is the ratio of djusted Monto ment over the	4 4 2 2 data are acco the area t e-Carlo res full range	4.07 ± 0.06 3.37 ± 0.06 2.03 ± 0.03 urate within abo o the close-pa sults show ver	4.34 3.59 2.12 out 1%. acked area.	1.700 1.800 2.000 3.000 Kinetic the a correction 12 partic	

5. Molecular Dynamics
 Potential Models Atomistic potential models, pair potentials, cutoff radius, three-body potential Reduced Units and the Principle of Corresponding States
Initial Conditions Positions, velocities, energy
 Boundary Conditions Fixed and Periodic Boundary Conditions for the One-Dimensional Chain Mirror Boundary Conditions, Moving Boundaries, Shear Boundaries
Normal Mode Solutions for the One Dimensional Chain 6-particle periodic chain 5-particle fixed boundary chain
Time Reversibility Levesque-Verlet Bit-Reversible Algorithm
Calculational Efficiency Neighbor Lists and Cell Model

The Development of Potential Models for Molecular Dynamics

The interactions which determine bulk properties are primarily electrostatic in nature. A quantum mechanical treatment is the most fundamental treatment of these interactions. Fortunately there are several approximations which provide very good potential models for classical dynamics.

- 1. Born-Oppenheimer approximation : The ion coordinates can be used as the basis for the wave function calculations because of the electron to proton mass ratio, 1/1836. Experiments with He-He interactions confirm this approximation.
- 2. Forces between molecules are much weaker than intramolecular forces (vibrations) so that some molecules can be treated as rigid molecules with the location specified with the center of mass coordinate of the molecule. This is not the case for polymers.
- 3. Because of the mass ratio we can use classical dynamics and statistical mechanics and add quantum corrections.
- 4. Another simplification arises because to a first approximation the intermolecular potential is additive in the number of interacting particles within the range of the potential.

$$\emptyset = \sum_{pairs} \emptyset(\mathbf{r}_{ij}) + \sum_{triples} \emptyset(\mathbf{r}_{ij}, \mathbf{r}_{ik}, \mathbf{r}_{jk}) + \cdots$$

J. A. Barker and . Henderson: "What is "Liquid"?, Rev. Mod. Phys. 48, (October 1976).





Atom	Source	$\epsilon/k_{\rm B}({\rm K})$	$\sigma(nm)$
Н	[Murad and Gubbins 1978]	8.6	0.281
He	[Maitland et al. 1981]	10.2	0.228
С	[Tildesley and Madden 1981]	51.2	0.335
N	[Cheung and Powles 1975]	37.3	0.331
0	[English and Venables 1974]	61.6	0.295
F	[Singer et al. 1977]	52.8	0.283
Ne	[Maitland et al. 1981]	47.0	0.272
S	[Tildesley and Madden, 1981]	183.0	0.352
Cl	[Singer et al. 1977]	173.5	0.335
Ar	[Maitland et al. 1981]	119.8	0.341
Br	[Singer et al. 1977]	257.2	0.354
Kr	[Maitland et al. 1981]	164.0	0.383

Model Potentials Characterizing Material Behavior

Embedded-Atom Potentials for Metals :

Experimental data provide parameters for potentials for specific materials. Models for material behavior can be approximated from these data. Electrons in a metal form a charge-neutralizing background for the motion of the ions. The potentials for our two-dimensional copper simulations were a Lennard-Jones 6-12 potential and a density-dependent attractive potential, $\phi_{EA} \sim (\rho - 1)^2/2$, modeling the electronic distribution.

Long-Range Electrostatic Potentials

The method of Ewald sums is a reliable technique for calculating the long range interactions without resorting to the N² computational effort in the sum over all of the particles. The method is based on Fourier series and applies to periodic boundaries only. Ewald sums and related methods are described in detail in Frenkel and Smit's book.

Potentials Fit to Experimental Data

Curve-fitted experimental data has resulted in libraries of equation of state data for large collections of materials used at the National Laboratories in the United States.

Daan Frenkel and Berend Smit, Understanding Molecular Simulation : From Algorithms to Applications, Academic Press, New York (2002).



Beyond Two Body Pair Potentials \rightarrow Three Body Potentials Three-body potentials are needed when three atoms are close enough to interact with each other. Quantum mechanical perturbation theory is used to compute three-body interactions between the electrons of the three atoms which have induced multipole interactions. The Axilrod-Teller potential, **Edward Teller** V_{iik}, is an example of dipole-dipole interactions. 1908-2003 $V_{ijk} = E_0 \left[\frac{1 + 3\cos(\gamma_i)\cos(\gamma_j)\cos(\gamma_k)}{\left(r_{ij}r_{jk}r_{ik}\right)^3} \right]$ $V_{ijk} = -E_0/4$ 1943 $\frac{1}{1} \quad V_{ijk} = -3E_0/8$ Where r_{ij} is the distance between the atoms I and j, and γ_i is the angle between the two vectors. The coefficient E_0 is positive and of the order $V\alpha^3$ where V is the ionization energy and α is the mean atomic polarizability. The exact value of E₀ depends on the magnitude of the dipole matrix elements and the energies of the p orbitals.

Axilrod, B. M.; Teller, E. (1943). "Interaction of the van der Waals Type Between Three Atoms", Journal of Chemical Physics, **11 (6): 299**.



Reduced Units for Short-Range Pair Potentials in Mechanical Systems with a Single Type of Particle

For systems with just one type of particle set m = 1 and a pair potential with well depth ϵ and particle diameter σ , the other variables can be expressed in terms of reduced units :

Density	$\rho * = \rho \sigma^3 \qquad \qquad$
Energy	$E^* = E/\varepsilon$ $\varphi_{LJ} = 4\epsilon \left \left(\frac{-}{r} \right) - \left(\frac{-}{r} \right) \right $
Temperature	$T^* = k_B T / \epsilon$
Pressure	$\mathbf{P}^* = \mathbf{P}\sigma^3/\varepsilon$
Time	$t^* = (\epsilon/m\sigma^3)^{1/2} t$
Force	$f^* = f\sigma/\epsilon$
Torque	$\tau^* = \tau / \varepsilon$
Surface Tension	$\gamma^{\star} = \gamma \sigma^2 / \epsilon$

Boltzmann's constant needs to be included to express the equivalence between the energy units of heat and work . When the parameters such as ϵ , σ , and k_bT are set to one, they need not be carried along in the computer program.

Reduced properties are used to define corresponding states.















One-Dimensional Chain Displacements and Forces in Matrix Form					
6 particle chain with periodic boundary conditions					
$\begin{pmatrix} -2 & 1 & 0 & 0 & 0 & 1 \\ 1 & -2 & 1 & 0 & 0 & 0 \\ 0 & 1 & -2 & 1 & 0 & 0 \\ 0 & 0 & 1 & -2 & 1 & 0 \\ 0 & 0 & 0 & 1 & -2 & 1 \\ 1 & 0 & 0 & 0 & 1 & -2 \end{pmatrix} \begin{pmatrix} x_1 \\ x_2 \\ x_3 \\ x_4 \\ x_5 \\ x_6 \end{pmatrix} = \begin{pmatrix} F_1 \\ F_2 \\ F_3 \\ F_4 \\ F_5 \\ F_6 \end{pmatrix}$	F1 = -2*q1 + q6 + q2 F2 = -2*q2 + q1 + q3 F3 = -2*q3 + q2 + q4 F4 = -2*q4 + q3 + q5 F5 = -2*q5 + q4 + q6 F6 = -2*q6 + q5 + q1				
5 particle chain with fixed boundary conditions					
$\begin{pmatrix} -2 & 1 & 0 & 0 & 0 \\ 1 & -2 & 1 & 0 & 0 \\ 0 & 1 & -2 & 1 & 0 \\ 0 & 0 & 1 & -2 & 1 \\ 0 & 0 & 0 & 1 & -2 & 1 \\ 0 & 0 & 0 & 1 & -2 \end{pmatrix} \begin{pmatrix} x_1 \\ x_2 \\ x_3 \\ x_4 \\ x_5 \end{pmatrix} = \begin{pmatrix} F_1 \\ F_2 \\ F_3 \\ F_4 \\ F_5 \end{pmatrix}$	F1 = + x2 - 2.0d00*x1 F2 = x3 + x1 - 2.0d00*x2 F3 = x4 + x2 - 2.0d00*x3 F4 = x5 + x3 - 2.0d00*x4 F5 = + x4 - 2.0d00*x5				

			Particle	Displace	ements		
/lode							
#	k	1	2	3	4	5	6
1	$1\pi/3$	+1/2	-1/2	-1	-1/2	+1/2	+1
2	$2\pi/3$	-1/2	-1/2	+1	-1/2	-1/2	+1
3	$3\pi/3$	-1	+1	-1	+1	-1	+1
4	$4\pi/3$	-1/2	-1/2	+1	-1/2	-1/2	+1
5	5π/3	+1/2	-1/2	-1	-1/2	+1/2	+1
6	6π/3	+1	+1	+1	+1	+1	+1
Norn	nal mod	e period:	$s: \tau_n =$	(2π , 2π	$\pi/\sqrt{3}$, $2\pi/2$	2,2 $\pi/\sqrt{3}$, 2π)

5 Particle Chain with fixed boundaries Normal Mode Displacements

Mode #	k	λ
1	1(2π/12)	12
2	2(2π/12)	6
3	3(2π/12)	4
4	4(2π/12)	3
5	5(2π/12)	5/12

Normal Mode Periods (12.138, 6.2832, 4.442, 3.6276, 3.2524) $\tau_n = \left\{ 2\pi/(2 - \sqrt{3})^{1/2}, 2\pi, 2\pi/\sqrt{2}, 2\pi/\sqrt{3}, 2\pi/((2 + \sqrt{3})^{1/2}) \right\}$

Calculated values for dt = 0.0001 and 63000 cycles

 $\tau_n=\ (12.\,139, 6.\,283, 4.\,45, 3.\,6276, 3.\,26\)$

